

NANO HIGHLIGHT

Wetting of Surfaces with Nano-Scale Topology

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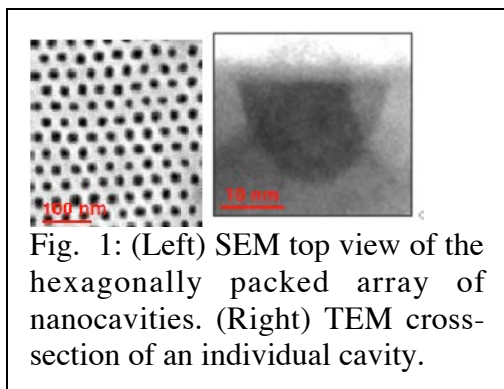


Fig. 1: (Left) SEM top view of the hexagonally packed array of nanocavities. (Right) TEM cross-section of an individual cavity.

confined to spaces that are not much larger than the molecular size are often different than those of unconfined liquids. This occurs because of the increased influence of the surface.

We have used synchrotron generated X-rays to study the manner in which liquids condense on nano-scale patterns formed on the surface of silicon by a combination of reactive ion etching and nano-scale patterning with block co-polymers, Fig. 1.[2] Condensation of a liquid, such as methyl-cyclohexane, is controlled by heating the silicon to a temperature ΔT higher than that of the liquid reservoir. The amount of condensation per unit area σ (i.e. electrons/ \AA^2), as determined from X-ray measurements, is shown by the blue/red curve in Fig. 2. The red line shows the temperature (T) dependence as the condensing liquid fills the cavities ($10\text{K} > \Delta T > 1\text{K}$). The blue line indicates the T -dependence for $\Delta T < 1\text{K}$ after the cavities are filled with liquid. The green line shows the results of the same experiment on a flat Si that has no nano-cavities. The T -dependence of condensation after the cavities have been filled ($\Delta T < 1\text{K}$) is identical to that of the flat Si; however, the growth following the initial adsorption is slower. These observations support recent theory.[3]

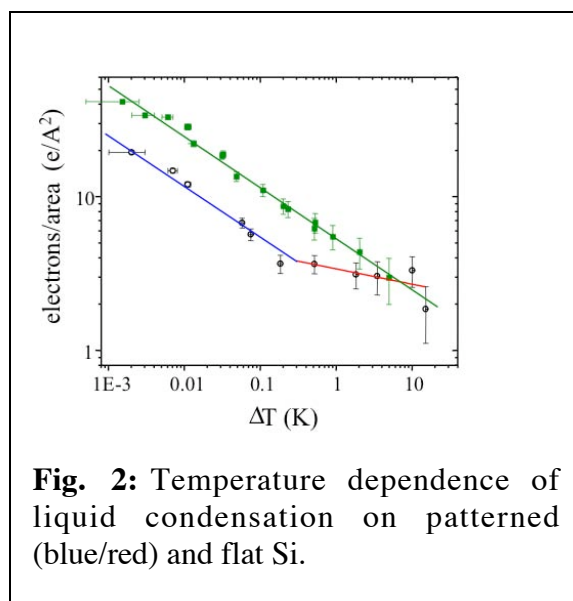


Fig. 2: Temperature dependence of liquid condensation on patterned (blue/red) and flat Si.

¹ C.P. Bacon, Y. Mattley and R. DeFrece, "Miniature spectroscopic instrumentation: Applications to biology and chemistry", Review of Scientific Instruments **75**, 1 (2004).

² T.P. Russell, T. Thurn-Albrecht, M. Tuominen, E. Huang and C.J. Hawker, "Block copolymers as nanoscopic templates", Macromolecular Symposia **159**, 77 (2000).

³ C. Rascon and A.O. Parry, "Surface phase diagrams for wetting on heterogenous substrates", Journal of Chemical Physics **115**, 5258 (2001).